

QUALI-START-UP SCIENCE LECTURE about IR-THERMOGRAPHY

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Motivation



Measurement of thermal radiation -> determination of temperature





Outline

- 1. Something about thermal radiation
- 2. Methods: passive and active thermography
- 3. Systems and laboratory at ZEA-1
- 4. Examples





Short history about IR-radiation

Around 1800: Discovery of infrared-radiation (Herrschel) 1860: Kirchhoff's definition of the "black body"

 $\alpha = \epsilon = 1, \quad (with \rho + \alpha + \tau = 1)$ 1879: Stefan–Boltzmann law – experimental (Stefan) 1884: Stefan–Boltzmann law – theoretical (Boltzmann) $P = \sigma \cdot A \cdot T^{4}, \ \sigma = \frac{2\pi^{5}k^{4}}{15h^{3}c^{2}} \approx 5,6704 \cdot 10^{-8} \frac{W}{m^{2}K^{4}} \text{ (stefan-Boltzmann constant)}$

1887: Proof: infrared-radiation = electromagnetic radiation (Hertz)1893: Wien's displacement law

$$\lambda_{max} = \frac{b}{T}$$
, $b \approx 2897,8 \ \mu m \ K$ (Wien's displacement constant)

1900: Planck's law

$$M(\lambda, T) = \frac{2\pi hc^2}{\lambda^5} \cdot \frac{1}{e^{\left(\frac{hc}{\lambda kT}\right)} - 1}$$





Spectrum of electromagnetic radiation



The IR-systems used at ZEA-1 operates both in sw and lw band





Wien's displacement law

black body radiation for different temperatures peaks at a wavelength inversely proportional to the temperature.







Planck's law

describes the spectral density of electromagnetic radiation emitted by a black body in thermal equilibrium at a given temperature T







Planck's law



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Example: red and yellow heat



Annealing colors for different



Some remarks about emissivity ϵ



Example: bodies with temperature of 100°C and different emissivities

Black body $\alpha = \epsilon = 1$

Every body in nature has an $\varepsilon < 1 \ ! \ 0,999999 > \varepsilon \ > 0,000001$





Some remarks about emissivity $\boldsymbol{\epsilon}$



Different emitters: Black body, gray body, selective emitter





$\varepsilon(\lambda)$ for different materials



metals:

in the lw-band only very small ϵ but we can increase ε by covering the surface (graphite spray, mylar foil,...) Rough surfaces, rusty surfaces or holes have a larger ε

in the sw-band ε is significantly larger

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nonmetals:

Have in general larger ϵ

e.g. human skin has $\varepsilon \approx 0.98$



Dependency of $\varepsilon(\lambda)$ from other parameterns

Surface:

in general: the rougher the surface, the larger $\boldsymbol{\epsilon}$

in holes: ϵ increases drilling a hole in a metal with depth 6 x larger then diameter ϵ increases from 0,02 -> 0,9

Temperature:

 ϵ can change due to phase transition increases with increasing temperature

metal	T [°C]	3
AI, uncoated	170	0,04
	500	0,05
AI, oxidized	200	0,11
	600	0,19
Stainless steel,	450	0,05
polished	500	0,065

Data from vdi-Wärmeatlas

All values of $\boldsymbol{\epsilon}$ in literature depend on the experimental conditions!





Dependency of $\varepsilon(\lambda)$ from the surface

Example: preheating plate for welding







Dependency of $\varepsilon(\lambda)$ from the surface

Metallic mirror on a heating plate, right side covered with graphite spray







Dependency of $\varepsilon(\lambda)$ from other parameters

Geometry or observation angle

metals: ϵ increases when the observation angle is increased nonmetals: ϵ decreases when the observation angle is increased



Source: itc-Germany





IR-Transmission of glasses



Glass 1737 - Alkaline Earth Boro-Aluminosilicate

most glasses (and plastics) are nontransparent in the lw-band





lw

10000

10% 21777nm

10000

SW

Passive and active thermography

Thermography: measuring the radiation emitted from the surface of an object

Thermography is a contactless method

to determine the real temperature of an object, one needs to know:

 ϵ , T_{Ambient}, τ , distance, relative humidity etc.

only useful in combination with conventionel measuring techniques like thermocouples or Pt_{100} -resistors but if one needs only a relative measurement (i.e. temperature changing)

passive thermography: measuring the emitted radiation and nothing else Applications:

- builling- ,plant- and production supervision
- thermal processes
- medical applications
- safety precautions, defense

active thermography: thermal excitation of the object and measuring simultaneously the emitted radiation Applications:

- Detection of defects like delamination, cracks, hidden structures,...
- Determination of material parametes like thermal diffusity, conductivity
- Nondestructive testing method





Passive thermography



Thermal relection at a fire door of two persons, which are standing the thermography camera SP01: 24,9°C SP02: 24,3°C





Active thermography

Pulse thermography



Lock-In thermography







Systems and laboratory at ZEA-1

sw-band: (two systems)

- Cooled (LN₂-Temperature) InSb-Detector with 640 x 512 IR-Pixel
- sensitive between 2,7 μ m 5,7 μ m
- sensitive (calibrated) between -20°C und 2000°C
- thermal resolution at 30°C: better then 0,018 K
- full frame mode 125 fps , subframe mode > 1000 fps
- full frame mode 350 fps, subframe mode > 5 kfps
- Several different object lenses (wide angle, tele, makro)
- Several filters to suppress stray radiation

Systems are mainly used for active thermography Not "transportable"





Systems and laboratory at ZEA-1

lw-band (7,5 μ m – 14 μ m)

- Microbolometer, 1024 x 738 IR-Pixel, uncooled
- calibrated -40°C 120°C, 0°C 500°C, 500°C 2000°C
- thermal resolution at 30°C: 0,03 K
- transportabel
- maximum full frame rate 60 (120) fps
- Several different object lenses (wide angle, tele, makro)

Mainly used for passive thermography Transportable





Systems and laboratory at ZEA-1

- Lock-In thermography, transient thermography, pulse thermography, flow thermography
- excitationssources like flash lamps, quartz radiator, eddy current generator, power supplies.
- TSA (Thermografic Stress Analysis
- Infrared calibrators
 - $25^{\circ}\text{C} 500^{\circ}\text{C}$, $\epsilon(\text{Iw}) = 0.93$; $\epsilon(\text{sw}) = 0.78$
 - 500°C 1500°C, ε(lw) = 0,98; ε(sw) = 0,98
 - Conventional thermal measurement equipment: thermo couples (Type K, S, B), Pt100,...
- Heat cabinets and furnaces





Examples for passive thermography: buildings







Examples for passive thermography: heat distribution during and after arc welding





Part of coil separators of the fusion reactor W7-X





Examples for passive thermography: heat distribution during and after arc welding



time lapse movie





Examples for passive thermography: heat distribution during and after arc welding



Thermo couple measurements





Examples for passive thermography: heat distribution during electron beam welding



sample: stainless steel



protective sheet for IR-window



sample mounted in the EB-vacuum chamber



bottom side: holes for thermo couples





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Examples for passive thermography: heat distribution during electron beam welding



After half of welding



2/3 of welding



2/3 post heat weld treatment Mitglied der Helmholtz-Gemeinschaft





Examples for passive thermography: heat distribution during electron beam welding



WF-0002: entlang Schweißnaht, 2/3 des Schweißprozesses 3000 2500 Messbereich: 500°C - 3000°C 2000 [°] scheinbare Temperaturerhöhung $-\epsilon = 0.30$ durch "Hohlraumeffekt" atur 1500 • $-\epsilon = 0.10$ $4 - \epsilon = 0.15$ Temper τ = 0.20
 ε = 0.25 1000 --ε = 0,93 500 . ich unterschritter 50 100 150 200 0 Abstand [mm] bezüglich Plattenanfang, horizontal

WF-0002: senkrecht zur Schweißnaht. 2/3 der Wärmenachbehandlur



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Examples for active thermography: short circuit in fuel cells









Examples for active thermography: short circuit in fuel cells







Examples for active thermography: short circuit in fuel cells







Examples for active thermography: gradient coils for MRT-spectrometers









Examples for active thermography: gradient coils for MRT-spectrometers



Three different coils G_x , G_y , G_z used current: I = 60 A, U = 4 V Movie is in time lapse mode





Examples for passive thermography: cooling systems for particle detectors (MVD for PANDA)







Examples for passive thermography: Dynamic evaluation of the cooling process

without cooling

beginning of cooling



with cooling







Analyzing the thermal behavior of a complex structure

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Analyzing the thermal behavior of a complex structure

Picture of a thermal wave going through a carbon carrier containing a cooling pipe



Calculated heat conductivity	
0,113 W/(m*K)	α = 1,115 E-07 m ² /K
0,117 W/(m*K)	α = 1,073 E-07 m ² /K
0,128 W/(m*K)	α = 1,216 E-07 m ² /K

estimated: density:1500 kg/m³ heat capacity: 700 J/(kg*K)





Résumé

- 1. thermography is a useful modern measuring technique
- 2. ZEA-1 has state of the art thermography systems, labs and knowledge
- 3. To perform thermography investigations it is necessary to know all parameters, esp. the emissivity

Thank you very much for your attention



